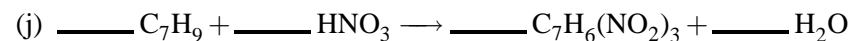
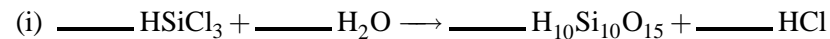
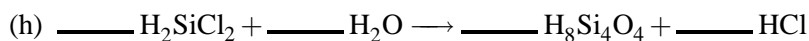
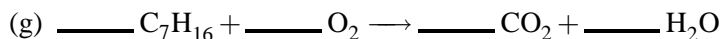
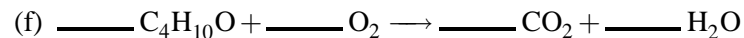
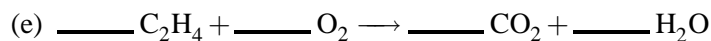
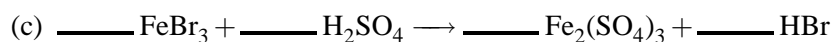


Balancing Equations: Practice Problems

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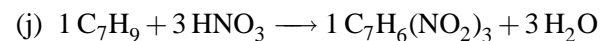
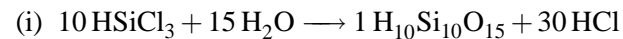
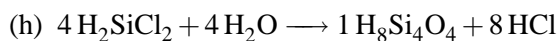
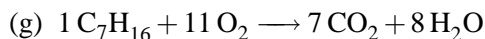
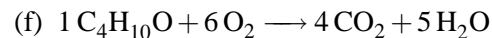
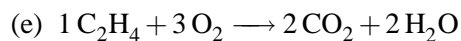
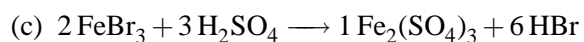
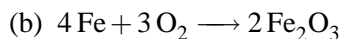
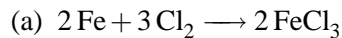
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1. Balance each of the following equations.



Balancing Equations: Answers to Practice Problems

1. Balanced equations. (Coefficients equal to one (1) do *not* need to be shown in your answers).



Name: _____

Date: _____

Balancing Equations

Balance the following chemical equations.

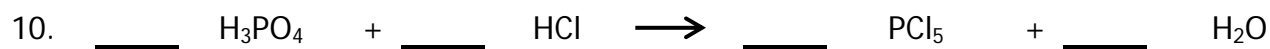
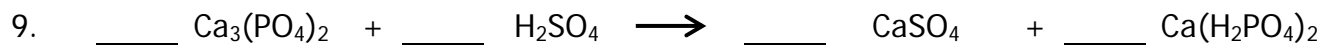
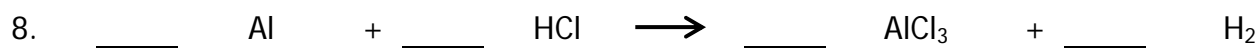
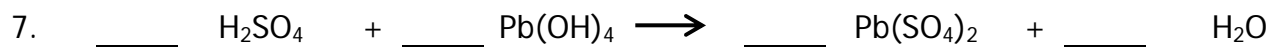
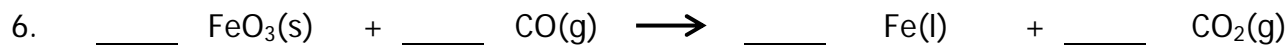
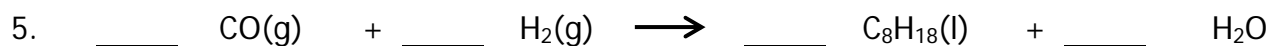
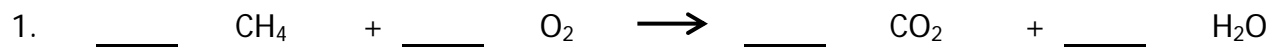
- _____ Fe + _____ H₂SO₄ → _____ Fe₂(SO₄)₃ + _____ H₂
- _____ C₂H₆ + _____ O₂ → _____ H₂O + _____ CO₂
- _____ KOH + _____ H₃PO₄ → _____ K₃PO₄ + _____ H₂O
- _____ SnO₂ + _____ H₂ → _____ Sn + _____ H₂O
- _____ NH₃ + _____ O₂ → _____ NO + _____ H₂O
- _____ KNO₃ + _____ H₂CO₃ → _____ K₂CO₃ + _____ HNO₃
- _____ B₂Br₆ + _____ HNO₃ → _____ B(NO₃)₃ + _____ HBr
- _____ BF₃ + _____ Li₂SO₃ → _____ B₂(SO₃)₃ + _____ LiF
- _____ (NH₄)₃PO₄ + _____ Pb(NO₃)₄ → _____ Pb₃(PO₄)₄ + _____ NH₄NO₃
- _____ SeCl₆ + _____ O₂ → _____ SeO₂ + _____ Cl₂

Name: _____

Date: _____

Balancing Equations

Balance the following chemical equations.

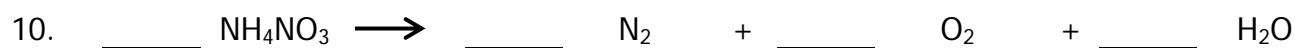
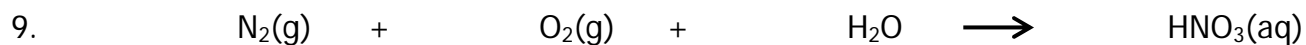
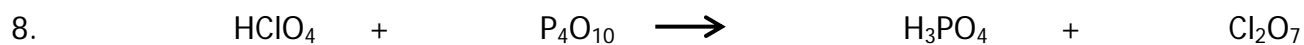
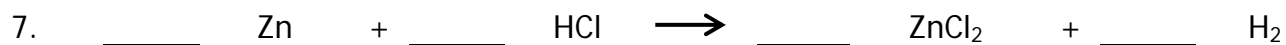
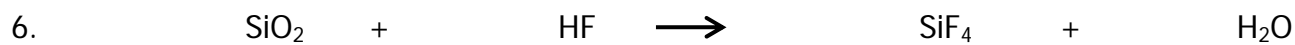
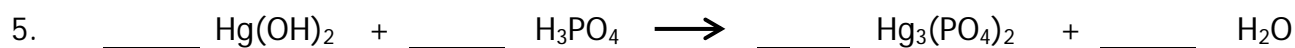
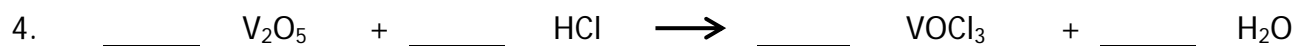
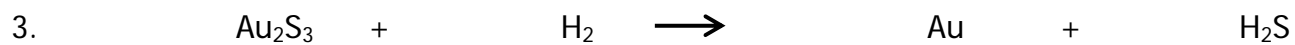
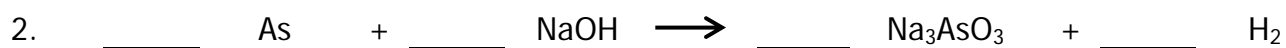
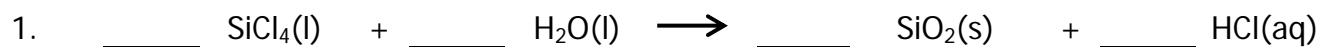


Name: _____

Date: _____

Balancing Equations

Balance the following chemical equations.

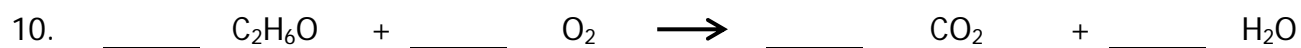
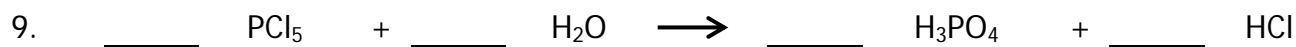
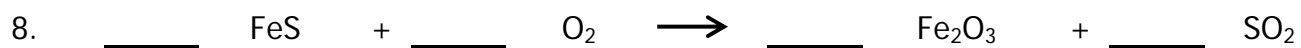
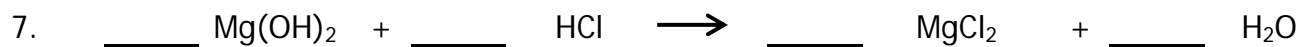
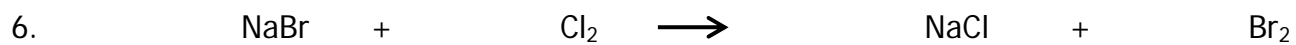
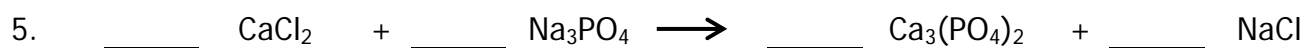
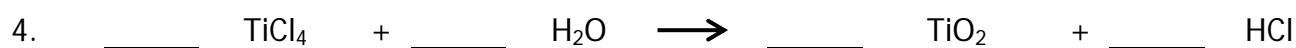
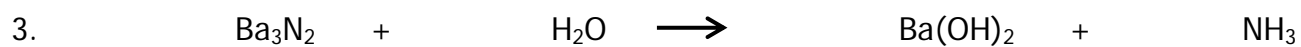
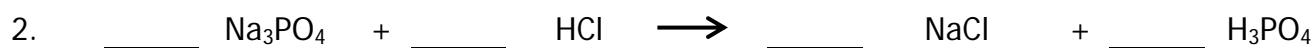
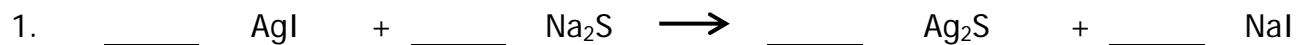


Name: _____

Date: _____

Balancing Equations

Balance the following chemical equations.



Name: _____

Date: _____

Balancing Equations

Balance the following chemical equations.

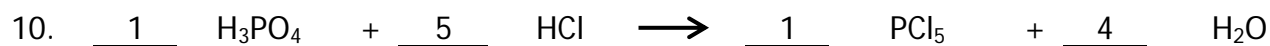
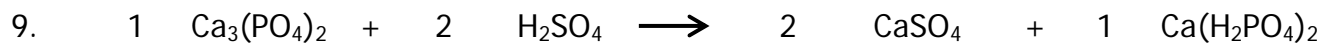
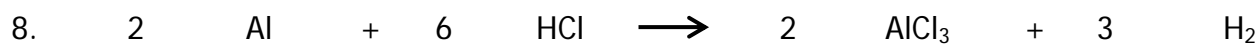
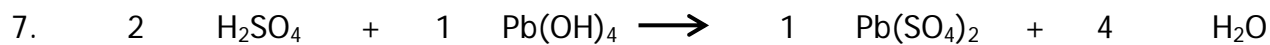
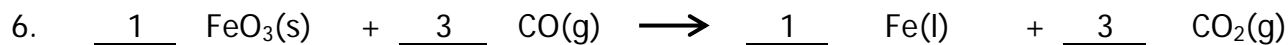
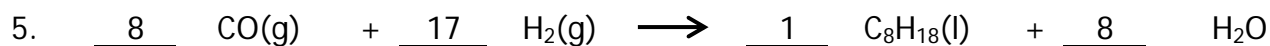
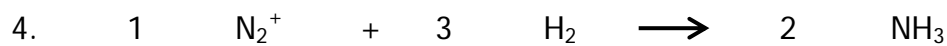
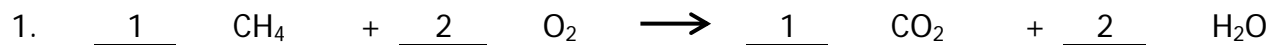
- $\underline{2} \text{ Fe} + \underline{3} \text{ H}_2\text{SO}_4 \longrightarrow \underline{1} \text{ Fe}_2(\text{SO}_4)_3 + \underline{3} \text{ H}_2$
- $\underline{2} \text{ C}_2\text{H}_6 + \underline{7} \text{ O}_2 \longrightarrow \underline{6} \text{ H}_2\text{O} + \underline{4} \text{ CO}_2$
- $\underline{3} \text{ KOH} + \underline{1} \text{ H}_3\text{PO}_4 \longrightarrow \underline{1} \text{ K}_3\text{PO}_4 + \underline{3} \text{ H}_2\text{O}$
- $\underline{1} \text{ SnO}_2 + \underline{2} \text{ H}_2 \longrightarrow \underline{1} \text{ Sn} + \underline{2} \text{ H}_2\text{O}$
- $\underline{4} \text{ NH}_3 + \underline{5} \text{ O}_2 \longrightarrow \underline{4} \text{ NO} + \underline{6} \text{ H}_2\text{O}$
- $\underline{2} \text{ KNO}_3 + \underline{1} \text{ H}_2\text{CO}_3 \longrightarrow \underline{1} \text{ K}_2\text{CO}_3 + \underline{2} \text{ HNO}_3$
- $\underline{1} \text{ B}_2\text{Br}_6 + \underline{6} \text{ HNO}_3 \longrightarrow \underline{2} \text{ B}(\text{NO}_3)_3 + \underline{6} \text{ HBr}$
- $\underline{2} \text{ BF}_3 + \underline{3} \text{ Li}_2\text{SO}_3 \longrightarrow \underline{1} \text{ B}_2(\text{SO}_3)_3 + \underline{6} \text{ LiF}$
- $\underline{4} \text{ (NH}_4)_3\text{PO}_4 + \underline{3} \text{ Pb}(\text{NO}_3)_4 \longrightarrow \underline{1} \text{ Pb}_3(\text{PO}_4)_4 + \underline{12} \text{ NH}_4\text{NO}_3$
- $\underline{1} \text{ SeCl}_6 + \underline{1} \text{ O}_2 \longrightarrow \underline{1} \text{ SeO}_2 + \underline{3} \text{ Cl}_2$

Name: _____

Date: _____

Balancing Equations

Balance the following chemical equations.

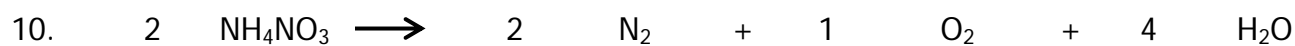
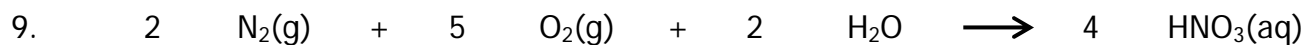
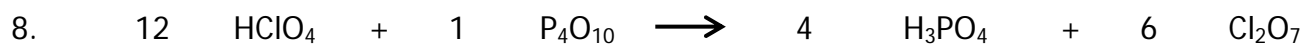
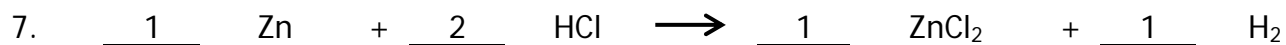
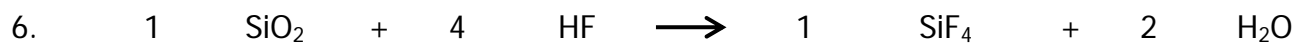
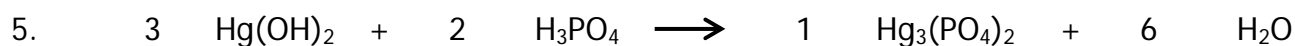
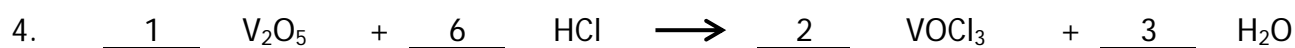
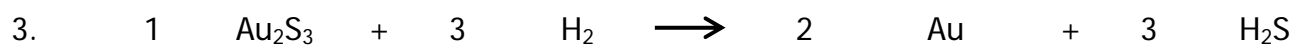
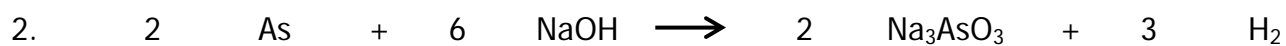
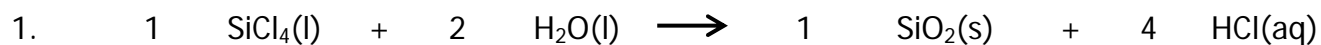


Name: _____

Date: _____

Balancing Equations

Balance the following chemical equations.



Name: _____

Date: _____

Balancing Equations

Balance the following chemical equations.

